This article was downloaded by:

On: 27 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

The Synthesis and Biological Evaluation of Regioisomeric Benzothiazolyl Coumarins

K. Shivashankar^a; Manohar V. Kulkarni^a; Lokesh A. Shastri^a; Vijaykumar P. Rasal^b; Sandur V. Rajendra^b Department of Postgraduate Studies in Chemistry, Karnatak University, Dharwad, India ^b College of Pharmacy, Karnataka Liberal Education Society, Hubli, India

To cite this Article Shivashankar, K. , Kulkarni, Manohar V. , Shastri, Lokesh A. , Rasal, Vijaykumar P. and Rajendra, Sandur V. (2006) 'The Synthesis and Biological Evaluation of Regioisomeric Benzothiazolyl Coumarins', Phosphorus, Sulfur, and Silicon and the Related Elements, 181:9,2187-2200

To link to this Article: DOI: 10.1080/10426500600614550 URL: http://dx.doi.org/10.1080/10426500600614550

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Phosphorus, Sulfur, and Silicon, 181:2187-2200, 2006

Copyright © Taylor & Francis Group, LLC ISSN: 1042-6507 print / 1563-5325 online

DOI: 10.1080/10426500600614550



The Synthesis and Biological Evaluation of Regioisomeric Benzothiazolyl Coumarins

K. Shivashankar Manohar V. Kulkarni Lokesh A. Shastri

Department of Postgraduate Studies in Chemistry, Karnatak University, Dharwad, India

Vijaykumar P. Rasal Sandur V. Rajendra

Karnataka Liberal Education Society, College of Pharmacy, Hubli, India

Various 4-aryloxymethylcoumarins have been obtained by the r.t. allylic substitution with formylphenols. These have been further reacted with o-aminothiophenol resulting in the formation of a benzothiazole skeleton. These compounds have been synthesised with a view to study their potential as microbial growth inhibitors. Comparative studies on the spectral and antimicrobial activities have also been carried out.

Keywords 4-Bromomethylcoumarins; antimicrobial; benzothiazole; coumarin

INTRODUCTION

The 1,3-benzothiazole moiety has been the seat of diverse biological properties through its innumerable derivatives.^{1–3} 2-arylbenzothiazoles have been found to be useful as topoisomerase inhibitors,^{4,5} in vivo imaging agents,⁶ lysophosphatidic acid acyltransfer agents,⁷ and prodrugs.⁸ The introduction of an aryloxy moiety in the pyran ring of coumarin leads to compounds with potential antitubercular⁹ and antimicrobial¹⁰ activities. Recently,

Received November 14, 2005; accepted December 30, 2005.

We thank the University Scientific Instrumentation Center–Karnatak University Dharwad, Indian Institute of Chemical Technology–Hyderabad, Indian Institute of Science–Bangalore, and Regional Sophisticated Instrumentation Centre–Chandigarh for providing spectral data. K. Shivashankar thanks University Grants Commission, New Delhi, for providing the FIP fellowship for his Ph.D.

Address correspondence to Manohar V. Kulkarni, Karnatak University, Department of Postgraduate Studies in Chemistry, Dharwad, 580003 India. E-mail: drmvk274@ yahoo.co.in

we have observed that a linking of biocompatible fragments like Vanillin and Paracetamol leads to molecular entities with enhanced antiinflammatory¹¹ and antimicrobial activities.¹² Further, benzothiazoles with a phenolic hydroxyl function have been isolated from marine sponges.¹³ The biodegradation of coumarin leads to a generation of similar polar groups.¹⁴ In view of this, the present article reports the synthesis of benzothiazolyl coumarins (Scheme 1) **5** and **8** via the formyl-4-aryloxy methyl coumarins **3** and **7**. The intermediates **3** and **7** possessing a formyl group have been reacted with *o*-aminothiophenol **4**, resulting in the formation of regioisomeric benzothiazolyl coumarins **5** and **8** respectively. The present synthetic strategy is illustrative of the 4+1 approach for the synthesis of five-membered heterocycles utilizing the double nucleophilicity of *o*-aminothiophenol and the electrophilic aldehydic carbon. All compounds have been subjected to a preliminary

SCHEME 1

antimicrobial screening against both Gram positive and Gram negative species.

The synthetic scheme for the target molecules was initiated by the Pechmann cyclization of phenols with 4-bromoethylacetoacetate¹⁵ leading to the required 4-bromomethylcoumarins⁹ 1. The r.t. allylic nucle-ophilic displacement was brought about by using formyl phenols 2 and 6, resulting in the formation of 4-aryloxymethyl coumarins 3 and 7. The so-obtained intermediates were refluxed with equimolar quantities of *o*-aminothiophenol in dimethylsulfoxide. The resulting high-melting solids separated in the reaction mixture and were filtered off to obtain compounds 5 and 8 as crystalline solids.

RESULTS AND DISCUSSION

Spectrum of 4-(2-oxo-2H-chromen-4-yl-methoxy)benzaldehyde 3a (R = 6-CH₃), the lactone carbonyl stretching frequency was observed at 1714 cm⁻¹, whereas the aldehydic carbonyl stretching appeared at 1696 cm⁻¹. In the ¹H NMR spectrum of compound 3a (R = 6-CH₃), a singlet was observed at δ 2.46 due to C₆-CH₃ protons. The C₄-CH₂ protons were observed downfield as singlet at δ 5.34. The C₃-H of coumarin appeared at δ 6.67. C₅-H, C_7 -H and C_8 -H of coumarin resonated as doublets at δ 7.33 ($J_{1,3} = 3.3$ Hz), δ 7.4 ($J_{1,2} = 8.7$ Hz, $J_{1,3} = 3.3$ Hz), and δ 7.27 ($J_{1,2} = 8.7$ Hz), respectively. Protons in the aryloxy moiety appeared as an AA'BB' pattern. Protons *ortho* to the –CHO group appeared as a doublet at δ 7.90 ($J_{1,2} = 8.4 \text{ Hz}$), and protons *ortho* to phenolic oxygen appeared as a doublet at δ 7.14 ($J_{1,2} = 8.4$ Hz). The aldehydic proton appeared as a singlet in the downfield at δ 9.95. The ¹³C NMR spectral data of compound 3a is given in the Experimental section, which is confirmed by its 2D-HETERO COSY spectrum. The assignments are in agreement with the literature reports on coumarin¹⁶ and benzaldehyde. ¹⁷

In the IR Spectrum of 3-(2-oxo-2H-chromen-4-yl-methoxy)-benzaldehyde 7a (R=6-CH₃), the lactone carbonyl stretching frequency was observed at 1707 cm⁻¹, whereas the aldehydic carbonyl stretching appeared at 1690 cm⁻¹. In the ¹H NMR spectrum of compound 7g (R = 6–Br), a singlet was observed at δ 5.52 due to C₄-CH₂ protons. The C₃-H of coumarin appeared at δ 6.66. The C₅-H of coumarin appeared as a singlet at δ 7.71. The C₇-H of coumarin resonated as doublets at δ 7.42 ($J_{1,2}$ = 8.7 Hz). The only proton in the aryloxy moiety flanked by a –CHO and –CH₂O– group appeared as a singlet at δ 8.13. The proton *ortho* to –CHO and *para* to the –CH₂O group appeared as a doublet at δ 7.81 ($J_{1,2}$ = 8.7 Hz). The C₈-H of coumarin and protons *meta* and *para* to the –CHO group

resonated as a multiplet in the region δ 7.59–7.60, which was too close for separation. The aldehydic proton appeared in the downfield as a singlet at δ 10.0 ppm.

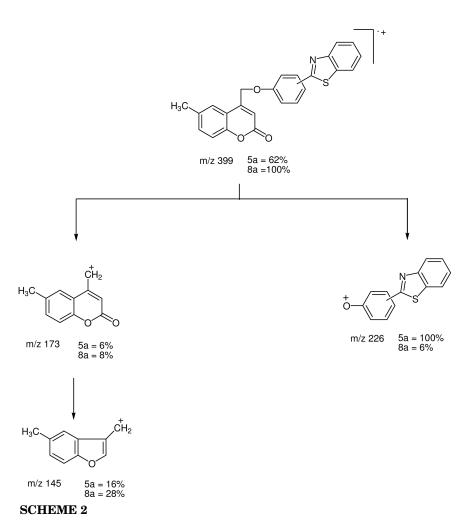
In the IR spectrum of 4-(4-benzothiazol-2-yl-phenoxymethyl)-chromen-2-one $\bf 5a$ (R = 6-CH₃), the lactone carbonyl stretching frequency was observed at 1715 cm⁻¹. The absence of a band around 1696 cm⁻¹ confirmed the conversion of aldehyde into the benzothiazole. In the ¹H NMR spectrum of compound $\bf 5a$ (R = 6-CH₃), a singlet was observed at δ 2.46 due to C₆-CH₃ protons. The C₄-CH₂ protons were observed downfield as a singlet at δ 5.4. The C₃-H of coumarin appeared at δ 6.66, whereas the aromatic protons resonated as a multiplet in the region δ 7.19–8.10. The absence of a –CHO proton at δ 9.95 was indicative of the formation of benzothiazole.

In the IR spectrum of 4-(3-benzothiazol-2-yl-phenoxymethyl)-chromen-2-one $\bf 8a$ (R = 6-CH₃), the lactone carbonyl stretching frequency was observed at 1718 cm⁻¹. The absence of a band around 1690 cm⁻¹ confirmed the conversion of aldehyde into the benzothiazole. In the ¹H NMR spectrum of compound $\bf 8a$ (R = 6-CH₃), a singlet was observed at δ 2.43 due to C₆-CH₃ protons. The C₄-CH₂ protons were observed downfield as a singlet at δ 5.43. The C₃-H of coumarin appeared at δ 6.61, whereas the aromatic protons resonated as a multiplet in the region δ 7.19–8.09. The absence of a —CHO proton at δ 10.0 confirmed the formation of benzothiazole.

The mass spectrum (EI) of compound $\mathbf{5a}$ (R = 6-CH₃) showed a molecular ion peak itself as a base peak at m/z 399 (100%). The mass spectrum (EI) of compound $\mathbf{8a}$ (R = 6-CH₃) showed a molecular ion peak at m/z 399 (62%), whereas a base peak appeared at m/z 226 (100%) due to homolytic allylic cleavage of the CH₂-O bond leading to the formation of aryloxy benzothiazole ion. The minor differences observed in the mass spectral fragmentation of the two-regio isomeric benzothiazoles is depicted in Scheme 2.

ANTIMICROBIAL ACTIVITY

The antimicrobial screening for all compounds was carried out against Gram positive and Gram negative species with *B. subtilis* and *E. coli*, respectively. *A. niger* and *C. albicans* were employed as fungal strains. DMF was used as a solvent control. The reference drugs used were *Ciprofloxacin* and *Gresiofulvin*. Tests were carried out by the cup plate method at a concentration of $100 \,\mu \mathrm{g} \,\mathrm{mL}^{-1}$. After 48 h of incubation at $37^{\circ}\mathrm{C}$, the zone of inhibition was measured in mm. The percent inhibition of test compounds was related to the standard whose zone of inhibition was taken as 100%. Among the formyl ethers 3e and 7e (R = 6—OCH₃),



3g and 7g (R = 6-Br) were found to show higher antimicrobial activity. The most effective compounds were 3f and 7f (R = 6-Cl), which were equally effective against the two bacterial and fungal species (Table I). Compounds with methyl and benzo substitution were less active.

A similar trend was observed in the benzothiazolyl coumarins $\bf 5$ and $\bf 8$, where $\bf 5e$ and $\bf 8e$ (R = 6-OCH₃) and $\bf 5g$ and $\bf 8g$ (R = 6-Br) showed percent inhibition in the range 78–88% specifically for $\it C.~albicans$. The highest activity against all strains was associated with chloro substitution and in particular, the compounds $\bf 5f$ and $\bf 8f$ showed growth inhibition of all the species, comparable to the standard (Table I).

TABLE I Results of Antimicrobial Assay

			•						
		$B.\ subtilis$		E. 0	$E.\ coli$	A.niger	ger	C.albicans	cans
Compound	R	$\begin{array}{c} Zone\ of\\ inhibition\ (mm) \end{array}$	Relative inhibition (%)		$\begin{tabular}{ll} Zone of & Relative \\ inhibition (mm) inhibition (\%) \end{tabular}$	Zone of inhibition (mm)	Relative inhibition (%)		Relative inhibition (%)
3a	6 -CH $_3$	16	55.5	18	9.99	19	72.2	19	7.77
3b	7-CH_3	16	55.5	18	9.99	19	72.2	20	72.2
3c	5,6-Benzo		50.0	17	61.1	17	61.1	19	72.2
3d	7,8-Benzo	0 15	50.0	17	61.1	17	61.1	18	9.99
3e	$6-0$ CH $_3$	17	61.1	19	72.2	20	7.77	20	77.7
3f	6-Cl	21	83.33	22	88.8	22	88.8	23	94.4
3g	6-Br	18	9.99	20	7.7.7	21	83.33	21	83.33
5a	6 -CH $_3$	17	61.1	16	55.5	17	61.6	20	77.7
5 b	7-CH_3	17	61.1	16	55.5	16	55.5	20	77.7
5c	5,6-Benzo) 15	50.0	15	20.0	16	55.5	19	72.2
2q	7,8-Benzo	0 15	50.0	15	20.0	15	50.0	18	9.99
5e	$6-0$ CH $_3$		9.99	18	9.99	18	9.99	21	83.33
J2	6-Cl	23	94.4	22	88.8	21	83.33	22	88.8
5g	6-Br	22	88.8	19	72.2	19	72.2	22	88.8
7a	6-CH_3	21	83.33	19	72.2	20	77.2	19	72.2
7b	7-CH_3	18	99.99	17	61.1	19	72.2	19	72.2
7c	5,6-Benze		99.99	17	61.1	19	72.2	18	9.99
7 d	7,8-Benze		99.99	17	61.1	19	72.2	18	9.99
7e	$6-0$ CH $_3$		88.8	20	7.7.7	21	83.33	21	83.33
JŁ	6-Cl	23	94.4	22	88.8	22	88.8	22	88.8
7g	6-Br	21	83.33	19	72.2	20	7.77	20	77.7
8 a	6-CH_3		9.99	19	72.2	19	72.2	20	77.7
98	7-CH_3		9.99	19	72.2	19	72.2	20	77.7
8c	5,6-Benze		61.1	18	9.99	18	9.99	19	72.2
p8	7,8-Benze		61.1	18	9.99	18	9.99	18	9.99
8e	$6-0$ CH $_3$		72.2	20	7.7.7	20	7.77	20	77.7
J8	6-Cl	22	88.8	23	94.4	22	88.8	22	88.8
8g	6-Br	20	7.77	22	88.8	20	7.77	22	88.8
DMF		9	I	9	I	9	I	9	I
Ciprofloxacin		24	100	24	100	I	I	I	I
Gresiofulvin		I	I	I	I	24	100	24	100

EXPERIMENTAL

Melting points were determined using an electric melting point apparatus (Shital scientific industries, Mumbai) and are uncorrected. IR spectra (KBr) were run on a Nicolet impact 410 FT-IR spectrometer ($\nu_{\rm max}$ in cm $^{-1}$). $^{1}{\rm H}$ NMR and $^{13}{\rm C}$ NMR spectra were recorded in CDCl3 and DMSO- d_{6} with TMS as an internal standard (chemical shift in δ ppm and J values in Hz) on a Brucker 300 MHz FTNMR spectrometer. A 2D-heterocosy spectrum was recorded on a 500 MHz NMR at the Indian Institute of Science, Bangalore. Elemental analyses were carried out on a Heraus CHN rapid analyzer. The purity of the compound was checked by TLC. Nomenclature was made using Chem. Draw ultra version 6.0. All reagents were of laboratory reagent quality, were purchased from SD.fine-Chem, and were used after purification.

The Preparation of (2-Oxo-2*H*-chromen-4-yl-methoxy)-benzaldehydes (3a–g) and (7a–g): General Procedure

Formylphenols **2** or **6** (1.22 g, 10 mmol) and anhydrous K_2CO_3 (1.38 g, 10 mmol) were stirred in dry acetone (50 mL) for 30 min. 4-bromomethyl coumarins (**1a–g**) (10 mmol) were added, and stirring was continued for 24 h. The reaction mixture was concentrated and poured into ice-cold water. The solid separated was filtered and washed with 5% HCl (10 mL) to neutralize the excess of potassium carbonate. Then it was washed with 100 mL of cold water and with ethanol. The crude product was dried and recrystallised from DMF.

4-(6-Methyl-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde (3a)

Colorless crystals from DMF. Yield 80%, m.p. 220–222°C; (found; C, 73.09; H, 4.29. $C_{18}H_{14}O_4$ (294.30) requires C, 73.46; H, 4.76%); IR: $\nu=1714$ (C=O, lactone), 1696 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta=2.46$ (s, 3H, C₆-CH₃), 5.34 (s, 2H, CH₂O), 6.67 (s, 1H, C₃-H), 7.14–7.93 (m, 7H, Ar–H), 9.95 (s, 1H, CHO) ppm; ¹³C NMR (CDCl₃) $\delta=161.0$ (C₂), 115.0 (C₃), 148.0 (C₄), 123.0 (C₅), 153.0 (C₆), 134.0 (C₇), 118.0 (C₈), 132.0 (C₉), 114.0 (C₁₀), 66.0 (C₁₁), 163.0 (C₁₂), 117.0 (C₁₃ & C₁₇), 133.0 (C₁₄ & C₁₆), 131.0 (C₁₅), 192.0 (C₁₈), and 21.0 (C₁₉) ppm.

4-(7-Methyl-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde (3b)

Colorless crystals from DMF. Yield 78%, m.p. 226–228°C; (found; C, 73.16; H, 4.31. $C_{18}H_{14}O_4$ (294.30) requires C, 73.46; H, 4.76%); IR: $\nu = 1701$ (C=O, lactone), 1689 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃):

 $\delta = 2.49 \; (s, 3H, C_6\text{-}CH_3), 5.32 \; (s, 2H, CH_2O), 6.61 \; (s, 1H, C_3\text{-}H), 7.14-7.92 \; (m, 7H, Ar-H), 9.94 \; (s, 1H, CHO) \; ppm.$

4-(3-Oxo-3H-benzo[f]chromen-1-ylmethoxy)-benzaldehyde) (3c)

Colorless crystals from DMF. Yield 80%, m.p. 222–224°C; (found; C, 76.09; H, 3.91. $C_{21}H_{14}O4$ (330.33) requires C, 76.36; H, 4.24%); IR: $\nu = 1710$ (C=O, lactone), 1696 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta = 5.73$ (s, 2H, CH₂O), 6.91 (s, 1H, C₃-H), 7.13–8.13 (m, 10H, Ar–H), 9.95 (s, 1H, CHO) ppm.

4-(2-Oxo-2H-benzo[h]chromen-4-ylmethoxy)-benzaldehyde) (3d)

Colorless crystals from DMF. Yield 75%, m.p. 246–248°C; (found; C, 73.04; H, 3.97. $C_{21}H_{14}O_4$ (330.33) requires C, 76.36; H, 4.24%); IR: $\nu = 1714$ (C=O, lactone), 1707 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta = 5.47$ (s, 2H, CH₂O), 6.74 (s, 1H, C₃-H), 7.13–8.67 (m, 10H, Ar–H), 9.93 (s, 1H, CHO) ppm.

4-(6-Methoxy-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde) (3e)

Colorless crystals from DMF. Yield 70%, m.p. 202–204°C; (found; C, 69.31; H, 4.24. $C_{18}H_{14}O_5$ (310.30) requires C, 69.67; H, 4.51%); IR: $\nu = 1712$ (C=O, lactone), 1702 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta = 3.88$ (s, 3H, C_6 -OCH₃), 5.31 (s, 2H, CH₂O), 6.68 (s, 1H, C_3 -H), 7.00–7.93 (m, 7H, Ar–H), 9.94 (s, 1H, CHO) ppm.

4-(6-Chloro-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde (3f)

Colorless crystals from DMF. Yield 66%, m.p. 220–222°C; (Found; C, 64.65; H, 3.17. $C_{17}H_{11}O_4Cl$ (314.72) requires C, 64.96; H, 3.50%); IR: $\nu=1720$ (C=O, lactone), 1700 (C=O, aldehyde) cm $^{-1}$; 1H NMR (CDCl $_3$): $\delta=5.28$ (s, 2H, CH $_2O$), 6.71 (s, 1H, C $_3$ -H), 7.12–7.93 (m, 7H, Ar–H), 9.90 (s, 1H, CHO) ppm.

4-(6-Bromo-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde (3g)

Colorless crystals from DMF. Yield 68%, m.p. 217–219°C; (found; C, 56.47; H, 2.81. $C_{17}H_{11}O_4Br$ (359.17) requires C, 56.82; H, 3.06%); IR: $\nu=1718$ (C=O, lactone), 1700 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta=5.54$ (s, 2H, CH₂O), 6.65 (s, 1H, C₃-H), 7.39–7.94 (m, 7H, Ar–H), 9.90 (s, 1H, CHO) ppm.

3-(6-Methyl-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde (7a)

Colorless crystals from DMF. Yield 72%, m.p. 202–204°C; (found; C, 73.11; H, 4.42. $C_{18}H_{14}O_4$ (294.30) requires C, 73.46; H, 4.76%); IR: $\nu=1707$ (C=O, lactone), 1690 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta=2.45$ (s, 3H, C_6 -CH₃), 5.29 (s, 2H, CH₂O), 6.67 (s, 1H, C_3 -H), 7.26–7.53 (m, 7H, Ar–H), 10.01 (s, 1H, CHO) ppm.

3-(7-Methyl-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde (7b)

Colorless crystals from DMF. Yield 70%, m.p. 223–225°C; (found; C, 73.12; H, 4.39. $C_{18}H_{14}O_4$ (294.30) requires C, 73.46; H, 4.76%); IR: $\nu = 1716$ (C=O, lactone), 1687 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta = 2.49$ (s, 3H, C₆-CH₃), 5.30 (s, 2H, CH₂O), 6.64 (s, 1H, C₃-H), 7.15–7.56 (m, 7H, Ar–H), 10.02 (s, 1H, CHO) ppm.

3-(3-Oxo-3H-benzo[f]chromen-1-ylmethoxy)-benzaldehyde) (7c)

Colorless crystals from DMF. Yield 78%, m.p. 168–168°C; (found; C, 76.12; H, 3.89. $C_{21}H_{14}O_4$ (330.33) requires C, 76.36; H, 4.24%); IR: $\nu=1724$ (C=O, lactone), 1697 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta=5.69$ (s, 2H, CH₂O), 6.92 (s, 1H, C₃-H), 7.33–8.14 (m, 10H, Ar–H), 10.01 (s, 1H, CHO) ppm.

3-(2-Oxo-2H-benzo[h]chromen-4-ylmethoxy)-benzaldehyde) (7d)

Colorless crystals from DMF. Yield 76%, m.p. 185–187°C; (found; C, 76.01; H, 3. 49. $C_{21}H_{14}O_4$ (330.33) requires C, 76.36; H, 4.24%); IR: $\nu=1721$ (C=O, lactone), 1684 (C=O, aldehyde) cm⁻¹; 1H NMR (CDCl₃): $\delta=5.72$ (s, 2H, CH₂O), 6.94 (s, 1H, C₃-H), 7.27–8.16 (m, 10H, Ar–H), 10.02 (s, 1H, CHO) ppm.

3-(6-Methoxy-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde) (7e)

Colorless crystals from DMF. Yield 70%, m.p. 280–282°C; (found; C, 69.31; H, 4.22. $C_{18}H_{14}O_5$ (310.30) requires C, 69.67; H, 4.51%); IR: $\nu=1713$ (C=O, lactone), 1688 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta=3.88$ (s, 3H, C_6 -OCH₃), 5.28 (s, 2H, CH₂O), 6.69 (s, 1H, C_3 -H), 6.99–7.58 (m, 7H, Ar–H), 10.01 (s, 1H, CHO) ppm.

3-(6-Chloro-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde (7f)

Colorless crystals from DMF. Yield 65%, m.p. 230–232°C; (found; C, 64.48; H, 3.17. $C_{17}H_{11}O_4Cl$ (314.72) requires C, 64.96; H, 3.50%); IR: $\nu=1723$ (C=O, lactone), 1681 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta=5.26$ (s, 2H, CH₂O), 6.72 (s, 1H, C₃-H), 7.26–7.59 (m, 7H, Ar–H), 10.01 (s, 1H, CHO) ppm.

3-(6-Bromo-2-oxo-2H-chromen-4-ylmethoxy)-benzaldehyde (7g)

Colorless crystals from DMF. Yield 64%, m.p. 224–226°C; (found; C, 56.43; H, 2.79. $C_{17}H_{11}O_4Br$ (359.17) requires C, 56.82; H, 3.06%); IR: $\nu = 1720$ (C=O, lactone), 1688 (C=O, aldehyde) cm⁻¹; ¹H NMR (CDCl₃): $\delta = 5.52$ (s, 2H, CH₂O), 6.66 (s, 1H, C₃-H), 7.42–8.13 (m, 7H, Ar–H), 10.00 (s, 1H, CHO) ppm.

The Preparation of 4-(Benzothiazol-2-yl-phenoxymethyl)-chromen-2-ones (5a-g) and (8a-g): General Procedure

(2-Oxo-2H-chromen-4-yl-methoxy)-benzaldehydes (**3a-g**) or (**7a-g**) (10 mmol) and o-aminothiophenol (1.1 mL, 10 mmol) were refluxed at 120^{0} C in an oil bath for 8 h in dimethylsulfoxide (10 mL). The reaction mixture was cooled to r.t., and the separated solid was filtered, washed with ethanol, dried and recrystallized from DMF.

4-(4-Benzothiazol-2-yl-phenoxymethyl)-6-methyl-chromen-2-one (5a)

Colorless crystals from DMF. Yield 92%, m.p. 300–302°C; (found; C, 71.86; H, 3.97; N, 3.19. $C_{24}H_{17}O_3NS$ (399.46) requires C, 72.18; H, 4.26; N, 3.5%); IR: $\nu=1715$ (C=O, lactone) cm $^{-1}$; 1H NMR (DMSO- d_6): $\delta=2.46$ (s, 3H, C_6 -CH $_3$), 5.40 (s, 2H, CH $_2$ O), 6.66 (s, 1H, C_3 -H), 7.19–8.10 (m, 11H, Ar–H) ppm; ms: m/z 399 (M+), m/z 371 (15%) (M–CO), m/z 384 (30%) (M–CH $_3$), m/z 356 (5%) (M–CO $_2$).

4-(4-Benzothiazol-2-yl-phenoxymethyl)-7-methyl-chromen-2-one (5b)

Colorless crystals from DMF. Yield 91%, m.p. 226–228°C; (found; C, 71.88; H, 3.94; N, 3.16. $C_{24}H_{17}O_3NS$ (399.46) requires C, 72.18; H, 4.26, N; 3.5%); IR: $\nu = 1708$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta = 2.46$ (s, 3H, C_6 -CH₃), 5.26 (s, 2H, CH₂O), 6.62 (s, 1H, C_3 -H), 7.07–8.08 (m, 11H, Ar–H) ppm.

1-(4-Benzothiazol-2-yl-phenoxymethyl)-benzo[f]chromen-3-one (5c)

Colorless crystals from DMF. Yield 91%, m.p. 262–264°C; (found; C, 74.16; H, 3.59; N, 2.91. $C_{27}H_{17}O_3NS$ (435.49) requires C, 74.48; H; 3.90; N, 3.21%); IR: $\nu=1708$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta=5.70$ (s, 2H, CH₂O), 6.95 (s, 1H, C₃-H), 7.11–8.16 (m, 14H, Ar–H) ppm.

4-(4-Benzothiazol-2-yl-phenoxymethyl)-benzo[h]chromen-2-one (5d)

Colorless crystals from DMF. Yield 93%, m.p. 140–142°C; (found; C, 74.16; H, 3.59; N, 2.91. $C_{27}H_{17}O_3NS$ (435.49) requires C, 74.48; H, 3.90; N, 3.21%); IR: $\nu=1715$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta=5.39$ (s, 2H, CH₂O), 6.99 (s, 1H, C₃-H), 7.04–8.03 (m, 14H, Ar–H) ppm.

4-(4-Benzothiazol-2-yl-phenoxymethyl)-6-methoxy-chromen-2-one (5e)

Colorless crystals from DMF. Yield 89%, m.p. 264–266°C; (found; C, 69.03; H, 3.71; N, 3.03. $C_{24}H_{17}O_4NS$ (415.46) requires C, 69.39; H, 4.09; N, 3.37%); IR: $\nu = 1718$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta = 3.84$ (s, 3H, C_6 -OCH₃), 5.45 (s, 2H, CH₂O), 6.60 (s, 1H, C_3 -H), 7.18–8.05 (m, 11H, Ar–H) ppm.

4-(4-Benzothiazol-2-yl-phenoxymethyl)-6-chloro-chromen-2-one (5f)

Colorless crystals from DMF. Yield 82%, m.p. 300–302°C; (found; C, 65.51; H, 3.01; N, 3.02. $C_{23}H_{14}O_3NSCl$ (419.88) requires C, 65.87; H, 3.34; N, 3.34%); IR: $\nu=1709$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta=5.38$ (s, 2H, CH₂O), 6.84 (s, 1H, C₃-H), 7.04–7.86 (m, 11H, Ar–H) ppm.

4-(4-Benzothiazol-2-yl-phenoxymethyl)-6-bromo-chromen-2-one (5g)

Colorless crystals from DMF. Yield 85%, m.p. 295–297°C; (found; C, 59.09; H, 2.69; N, 2.67. $C_{23}H_{14}O_3NSBr$ (464.33) requires C, 59.48; H, 3.01; N, 3.01%); IR: $\nu=1711$ (C=O, lactone) cm $^{-1}$; 1H NMR (DMSO- d_6): $\delta=5.36$ (s, 2H, CH $_2O$), 6.74 (s, 1H, C $_3$ -H), 7.19–8.10 (m, 11H, Ar–H) ppm.

4-(3-Benzothiazol-2-yl-phenoxymethyl)-6-methyl-chromen-2-one (8a)

Colorless crystals from DMF. Yield 91%, m.p. 225–227°C; (found; C, 71.82; H, 3.93; N, 3.15. $C_{24}H_{17}O_3NS$ (399.46) requires C, 72.18; H, 4.26; N, 3.5%); IR: $\nu=1718$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta=2.43$ (s, 3H, C_6 -CH₃), 5.43 (s, 2H, CH₂O), 6.61 (s, 1H, C_3 -H), 7.19–8.09 (m, 11H, Ar–H) ppm; ms m/z 399 (M⁺); m/z 371(3%) (M–CO), m/z 384 (2%) (M–CH₃).

4-(3-Benzothiazol-2-yl-phenoxymethyl)-7-methyl-chromen-2-one (8b)

Colorless crystals from DMF. Yield 90%, m.p. 214–216°C; (found; C, 71.79; H, 3.89; N, 3.11. $C_{24}H_{17}O_3NS$ (399.46) requires C, 72.18; H, 4.26; N; 3.5%); IR: $\nu = 1721$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta = 2.43$ (s, 3H, C_6 -CH₃), 5.31 (s, 2H, CH₂O), 6.66 (s, 1H, C_3 -H), 7.11–8.09 (m, 11H, Ar–H) ppm.

1-(3-Benzothiazol-2-yl-phenoxymethyl)-benzo[f]chromen-3-one (8c)

Colorless crystals from DMF. Yield 83%, m.p. 200–202°C; (found; C, 74.08; H, 3.51; N, 2.86. $C_{27}H_{17}O_3NS$ (435.49) requires C, 74.48; H; 3.90; N, 3.21%); IR: $\nu=1717$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta=5.71$ (s, 2H, CH₂O) 6.96 (s, 1H, C₃-H), 7.13–8.21 (m, 14H, Ar–H) ppm.

4-(3-Benzothiazol-2-yl-phenoxymethyl)-benzo[h]chromen-2-one (8d)

Colorless crystals from DMF. Yield 93%, m.p. 180–182°C; (found; C, 74.07; H, 3.49; N, 2.87. $C_{27}H_{17}O_3NS$ (435.49) requires C, 74.48; H; 3.90; N, 3.21%); IR: $\nu=1711$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta=5.42$ (s, 2H, CH₂O), 6.76 (s, 1H, C₃-H), 7.13–8.57 (m, 14H, Ar–H) ppm.

4-(3-Benzothiazol-2-yl-phenoxymethyl)-6-methoxy-chromen-2-one (8e)

Colorless crystals from DMF. Yield 89%, m.p. 206–208°C; (found; C, 69.02; H, 3.76; N, 3.01. $C_{24}H_{17}O_4NS$ (415.46) requires C, 69.39; H, 4.09; N, 3.37%); IR: $\nu = 1715$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta = 3.87$ (s, 3H, C_6 -OCH₃), 5.31 (s, 2H, CH₂O), 6.73 (s, 1H, C_3 -H), 7.04–8.09 (m, 11H, Ar–H) ppm.

4-(3-Benzothiazol-2-yl-phenoxymethyl)-6-chloro-chromen-2-one (8f)

Colorless crystals from DMF. Yield 81%, m.p. 180–182°C; (found; C, 65.51; H, 3.02; N, 3.03. $C_{23}H_{14}O_3NSCl$ (419.88) requires C, 65.87; H, 3.34; N, 3.34%); IR: $\nu=1715$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta=5.39$ (s, 2H, CH₂O), 6.77 (s, 1H, C₃-H), 7.15–8.58 (m, 11H, Ar–H) ppm.

4-(3-Benzothiazol-2-yl-phenoxymethyl)-6-bromo-chromen-2-one (8g)

Colorless crystals from DMF. Yield 84%, m.p. 230–232°C; (found; C, 59.48; H, 2.66; N, 2.69. $C_{23}H_{14}O_3NSBr$ (464.33) requires C, 59.48; H, 3.01; N, 3.01%); IR: $\nu=1721$ (C=O, lactone) cm⁻¹; ¹H NMR (DMSO- d_6): $\delta=5.42$ (s, 2H, CH₂O), 6.76 (s, 1H, C₃-H), 7.25–8.07 (m, 11H, Ar–H) ppm.

CONCLUSION

The 4-(benzothiazol-2-yl-phenoxymethyl)-chromen-2-ones **5** and **8** are more potent than formylethers **3** and **7**. Among regioisomers, the *meta* isomers are more potent than the *para* isomers both in the case of formylethers and benzothiazolyl coumarins. The compounds thatcontain the chloro, bromo, and methoxy substitution at the 6-position in the coumarin ring enhanced the growth inhibition in the following order: methoxy < bromo < chloro.

REFERENCES

- [1] T. D. Bradshaw and A. D. Westwell, Curr. Med. Chem., 11, 1009 (2004).
- [2] K. Akiba and N. Inamoto, Heterocycles, 7, 1131 (1977).
- [3] M. Yoshida, I. Hayakawa, N. Hayashi, T. Agatsuma, Y. Oda, F. Tanzawa, S. Iwasaki, K. Kouama, H. Furukawa, S. Kurakata, and Y. Sugano, *Bioorg. Med. Chem. Lett.*, 15, 3328 (2005).
- [4] A. D. Westwell, Drug Discov. Today, 7, 528 (2002).
- [5] J. S. Kim, Q. Sun, B. Gatto, C. Yu, A. Liu, L. F. Liu, and E. J. Lavoie, *Bioorg. Med. Chem.*, 4, 621 (1996).
- [6] D. Alagille, R. M. Baldwin, and G. D. Tamagnam, Tetrahedron Lett., 46, 1349 (2005).
- [7] B. Gong, F. Hong, C. Kohm, L. Bonham, and P. Klein, *Bioorg. Med. Chem. Lett.*, 14, 1455 (2004).
- [8] D. F. Shi, T. D. Bradshaw, M. Chua, A. D. Westwell, and M. F. G. Stevens, *Bioorg. Med. Chem. Lett.*, 11, 1093 (2001).
- [9] J. R. Merchant and A. S. Gupta, *Indian J. Chem.*, **22**, 158 (1979).
- [10] M. V. Kulkarni and V. D. Patil, Arch Pharm., 314, 708 (1981).
- [11] M. D. Ghate, M. V. Kulkarni, R. Shobha, and S. Y. Kattimani, Eur. J. Med. Chem., 38, 297 (2003).

- [12] L. A. Shastri, M. D. Ghate, and M. V. Kulkarni, Indian J. Chem., 43B, 2416 (2004).
- [13] A. A. Stierle, J. H. CardellinaII, and F. L. Singleton, Tetrahedron Lett., 32, 4847 (1991).
- [14] D. Egan, R. O'Kennedy, and E. Moran, Drug Metab. Rev., 22, 503 (1990).
- [15] A. Burger and G. E. Ullyot, J. Org. Chem., 12, 346 (1947).
- [16] W. V. Turner and W. H. Pirkle, J. Org. Chem., 39, 1935 (1974).
- [17] G. C. Levy and G. L. Nelson, Text Book of ¹³C NMR Spectroscopy, 2nd ed. (London: Heyden, 1976).
- [18] W. B. Hugo and A. B. Russel, in *Pharmaceutical Microbiology*, 4th ed., S. P. Denyer, N. A. Hodges, and S. P. Gorman, Eds., p. 187–202 (London: Blackwell Scientific Publications, 1987).